Unit 5 Slides

Molar mass, percent composition, and molar conversions

Bellwork on goformative

Objective:

SWBAT use values specific to chemical processes IOT calculate the mole, molar mass, molarity, and % composition.

Q: How rich is Avogadro?

A: He's a multi-mole-ionaire.

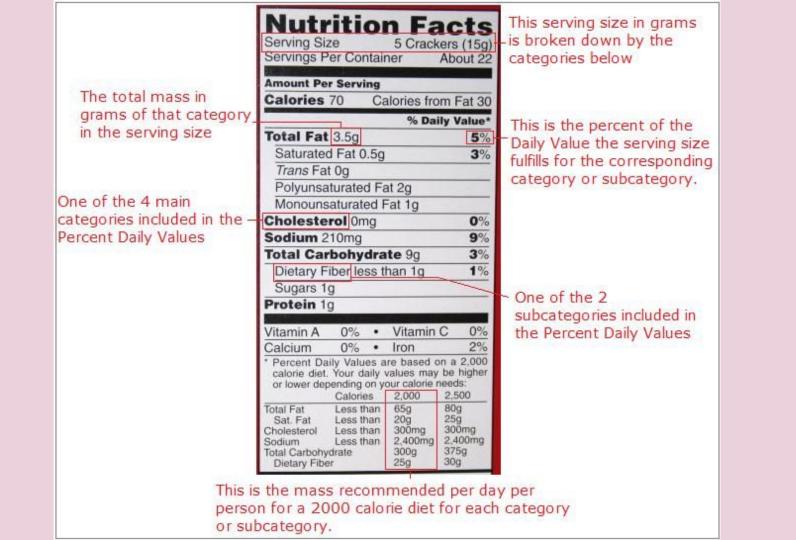


- Formative
- Agenda
- Objective
- Molar Mass Practice
 - CFU: Notes practice problems
- Mass Percent Gum Lab
 - CFU: Lab Worksheet
- Exit Ticket
 - On Formative

Turn and Talk: Which snack is healthier?

	about 15 servings per container Serving size 2 cookies (29g)		
Amount per serving Calories 1	40		
% Da	ily Value		
Total Fat 7g	9%		
Saturated Fat 2g	10%		
Trans Fat 0g			
Cholesterol Omg	0%		
Sodium 85mg	4%		
Total Carbohydrate 21g	8%		
Dietary Fiber Less than 1g	2%		
Total Sugars 13g			
Includes 13g Added Sugars	26%		
Protein Less than 1g			
Vitamin D Omeg	0%		
Calcium 9mg	0%		
Iron 0.92mg	6%		
Potassium 35mg	0%		

Nutriti	OI	n Facts
10 servings per c	onta	ainer
Serving size		1 Pastry (48g
Amount per serving Calories		170
SIT		% Daily Value
Total Fat 3g		49
Saturated Fat 1g		59
Trans Fat 0g	SOF	
Polyunsaturated Fa	at 1g	wist.
Monounsaturated I	Fat 0	.5g
Cholesterol 0mg		09
Sodium 120mg		59
Total Carbohydra	ite 3	37g 13 9
Dietary Fiber 3g		109
Total Sugars 15g		
Includes 15g Add	ded S	Sugars 309
Protein 2g		
Vitamin D 0mcg 0%		Calcium 130mg 109
Iron 1.8mg 10%	2005	Potassium 70mg 0°
Vitamin A 10%	•	Thiamin 10°
Riboflavin 10%	•	Niacin 10°
Vitamin B ₆ 10%		Hiddil 10



Molar Mass

Sum of the atomic masses

Units: g/mol

Allows us to convert between grams and moles

What is the molar mass of MgCl₂

Practice- What is the molar mass of

- 1. HNO₃
- 2. NaCl
- 3. Ca(OH)2
- 4. Al₂(SO₄)₃
- 5. C₆H₁₂O₆

Challenge problem: How many atoms are in a 3.56 g sample of Cu?

Percent composition

% composition= mass of a single element/mass of the whole compound x100



Work with your group to determine a procedure for determining the percentage of your gum that is made of sugar and the percentage that is made up of gum.

Hints for helping you come up with your procedure.

- Make sure you measure the mass of your gum BEFORE CHEWING (including the wrapper)
- Then make sure you measure the mass of your gum AFTER CHEWING (include the wrapper).
- As you chew the sugar dissolves and is no longer part of the gum. Thus sugar is removed in the chewing process.

EARLY FINISHER Unit 17-Work ModeEmpirical Formula from Mass Composition



Exit ticket

- 1. What is the molar mass of CsF?
- 2. Calculate the percent by mass of each element in Cesium Fluoride (CsF)?
- 3. A piece of gum is 34 g. If 10 g of that piece of gum is sugar, calculate the mass percentage that is made up of gum and the mass percentage that is made up of sugar.

Molar Conversions

https://drive.google.com/file/d/1AT1e25tzsdq4jRUKmweY-dJu5UkUprGc/view?usp=sharing