

Unit 5 Slides

Molar mass, percent composition, and molar conversions

Bellwork on goformative

Objective:

SWBAT **use** values specific to chemical processes IOT **calculate** the **mole, molar mass, molarity, and % composition.**

Q: How rich is Avogadro?

A: He's a multi-mole-ionaire.



- Formative
- Agenda
- Objective
- Molar Mass Practice
 - CFU: Notes practice problems
- Mass Percent Gum Lab
 - CFU: Lab Worksheet
- Exit Ticket
 - On Formative

Turn and Talk: Which snack is healthier?

Nutrition Facts	
about 15 servings per container	
Serving size 2 cookies (29g)	
Amount per serving	
Calories	140
% Daily Value*	
Total Fat 7g	9%
Saturated Fat 2g	10%
Trans Fat 0g	
Cholesterol 0mg	0%
Sodium 85mg	4%
Total Carbohydrate 21g	8%
Dietary Fiber Less than 1g	2%
Total Sugars 13g	
Includes 13g Added Sugars	26%
Protein Less than 1g	
Vitamin D 0mcg 0%	
Calcium 9mg 0%	
Iron 0.92mg 6%	
Potassium 35mg 0%	
* The % Daily Value (DV) tells you how much a nutrient in a serving of food contributes to a daily diet. 2,000 calories a day is used for general nutrition advice.	

Nutrition Facts	
10 servings per container	
Serving size	1 Pastry (48g)
Amount per serving	
Calories	170
% Daily Value*	
Total Fat 3g	4%
Saturated Fat 1g	5%
Trans Fat 0g	
Polyunsaturated Fat 1g	
Monounsaturated Fat 0.5g	
Cholesterol 0mg	0%
Sodium 120mg	5%
Total Carbohydrate 37g	13%
Dietary Fiber 3g	10%
Total Sugars 15g	
Includes 15g Added Sugars	30%
Protein 2g	
Vitamin D 0mcg 0%	
Calcium 130mg 10%	
Iron 1.8mg 10%	
Potassium 70mg 0%	
Vitamin A 10%	
Thiamin 10%	
Riboflavin 10%	
Niacin 10%	
Vitamin B₆ 10%	
* The % Daily Value (DV) tells you how much a nutrient in a serving of food contributes to a daily diet. 2,000 calories a day is used for general nutrition advice.	

Nutrition Facts

Serving Size 5 Crackers (15g)
 Servings Per Container About 22

Amount Per Serving

Calories 70 Calories from Fat 30

% Daily Value*

Total Fat 3.5g **5%**

Saturated Fat 0.5g **3%**

Trans Fat 0g

Polyunsaturated Fat 2g

Monounsaturated Fat 1g

Cholesterol 0mg **0%**

Sodium 210mg **9%**

Total Carbohydrate 9g **3%**

Dietary Fiber less than 1g **1%**

Sugars 1g

Protein 1g

Vitamin A 0% • Vitamin C 0%

Calcium 0% • Iron 2%

* Percent Daily Values are based on a 2,000 calorie diet. Your daily values may be higher or lower depending on your calorie needs:

	Calories	2,000	2,500
Total Fat	Less than	65g	80g
Sat. Fat	Less than	20g	25g
Cholesterol	Less than	300mg	300mg
Sodium	Less than	2,400mg	2,400mg
Total Carbohydrate		300g	375g
Dietary Fiber		25g	30g

This serving size in grams is broken down by the categories below

The total mass in grams of that category in the serving size

This is the percent of the Daily Value the serving size fulfills for the corresponding category or subcategory.

One of the 4 main categories included in the Percent Daily Values

One of the 2 subcategories included in the Percent Daily Values

This is the mass recommended per day per person for a 2000 calorie diet for each category or subcategory.

Molar Mass

Sum of the atomic masses

Units: g/mol

Allows us to convert between grams and moles

What is the molar mass of MgCl_2

Practice- What is the molar mass of

1. HNO_3
2. NaCl
3. $\text{Ca}(\text{OH})_2$
4. $\text{Al}_2(\text{SO}_4)_3$
5. $\text{C}_6\text{H}_{12}\text{O}_6$

Challenge problem: How many atoms are in a 3.56 g sample of Cu?

Percent composition

% composition = $\frac{\text{mass of a single element}}{\text{mass of the whole compound}} \times 100$



Work with your group to determine a procedure for determining the percentage of your gum that is made of sugar and the percentage that is made up of gum.

Hints for helping you come up with your procedure.

- Make sure you measure the mass of your gum BEFORE CHEWING (including the wrapper)
- Then make sure you measure the mass of your gum AFTER CHEWING (include the wrapper).
- As you chew the sugar dissolves and is no longer part of the gum. Thus sugar is removed in the chewing process.

EARLY FINISHER

Unit 17-Work Mode- Empirical Formula from Mass Composition



Exit ticket

1. What is the molar mass of CsF ?
2. Calculate the percent by mass of each element in Cesium Fluoride (CsF)?
3. A piece of gum is 34 g. If 10 g of that piece of gum is sugar, calculate the mass percentage that is made up of gum and the mass percentage that is made up of sugar.

Molar Conversions

<https://drive.google.com/file/d/1AT1e25tzsdq4jRUKmweY-dJu5UkUprGc/view?usp=sharing>